



Battery research for smart mobility: Achievements and new directions

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Electric mobility: an introduction

Battery research is business-driven
so are the basic research needs



by 2020
Today

300 Wh kg_{cell}⁻¹
Specific energy

650 Wh L_{cell}⁻¹
Energy density

95 % capacity at 1C
Power density

Sustainability
No cobalt

Cost
100 € kWh_{pack}⁻¹

Energy efficiency
98 %



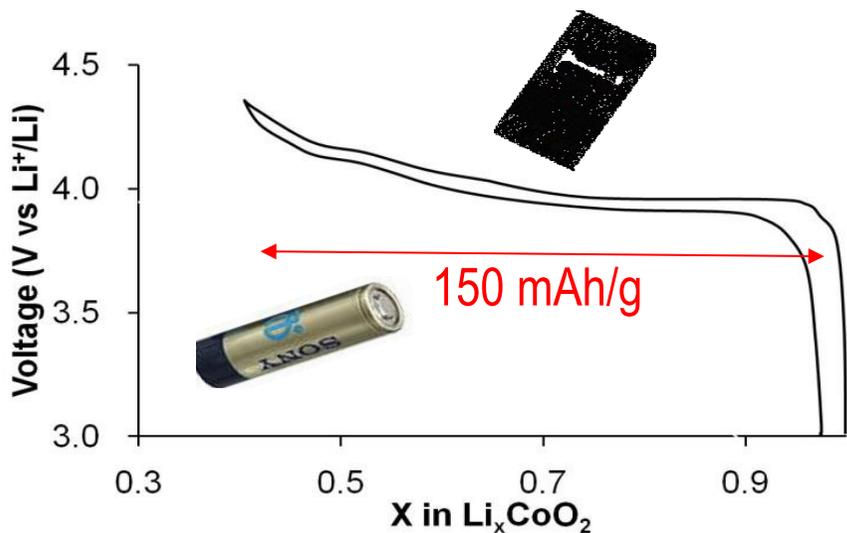
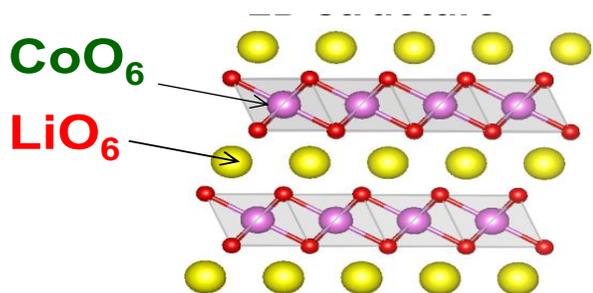
Source: Bloomberg New Energy Finance Scenario 5: Tesla Gigafactory NCA/graphite-Si

Revolution in the word of energy management
but also in the way that research must proceed



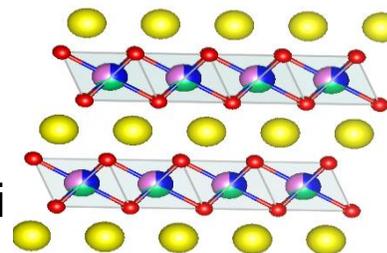
The layered oxides and their evolution through the years

LiCoO₂ (1991)

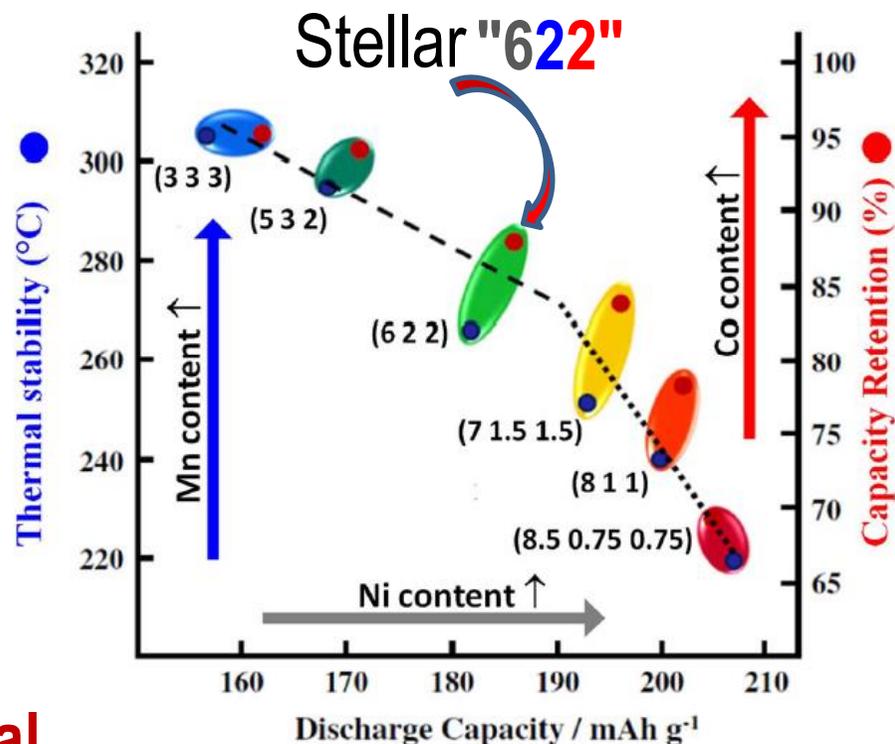


Chemically substituted samples

Replace partially Co by Mn and Ni



Li[NiMnCo]O₂
NMC phases
(170-1900 mAh/g)

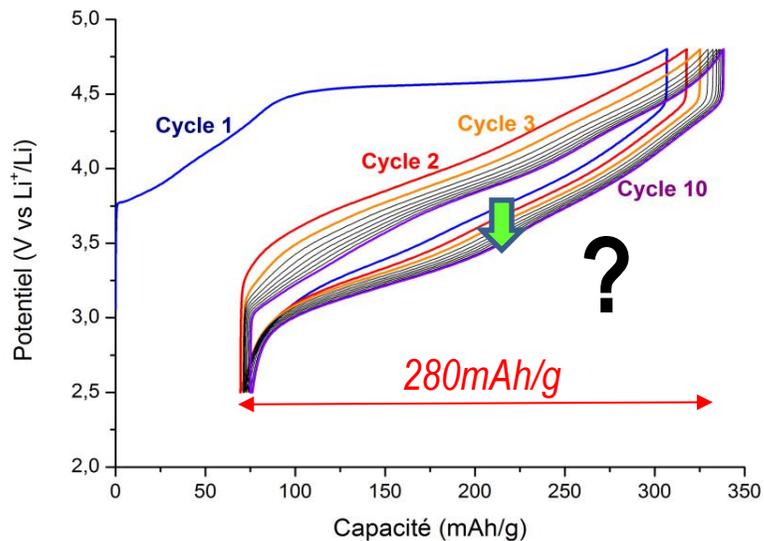
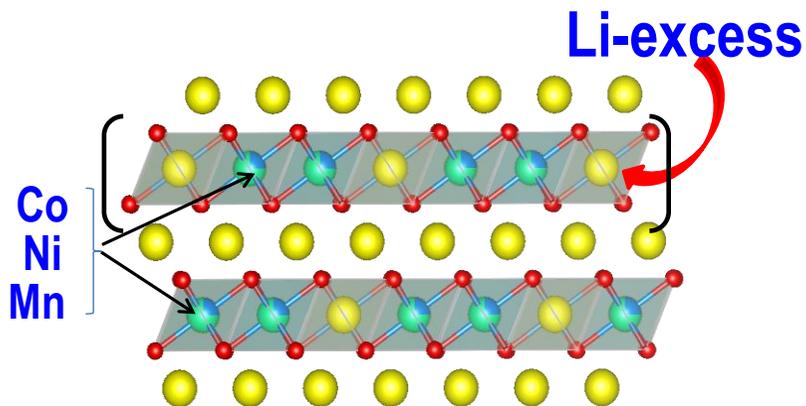


➔ Less than 1 Li⁺ exchanged per metal



Increasing the capacity further via chemical substitution

The Li-rich NMC phases



Design model materials

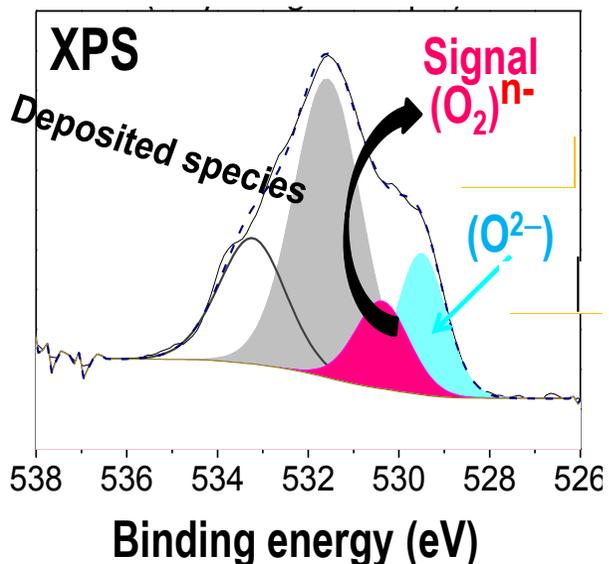
Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	3d
Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	4d



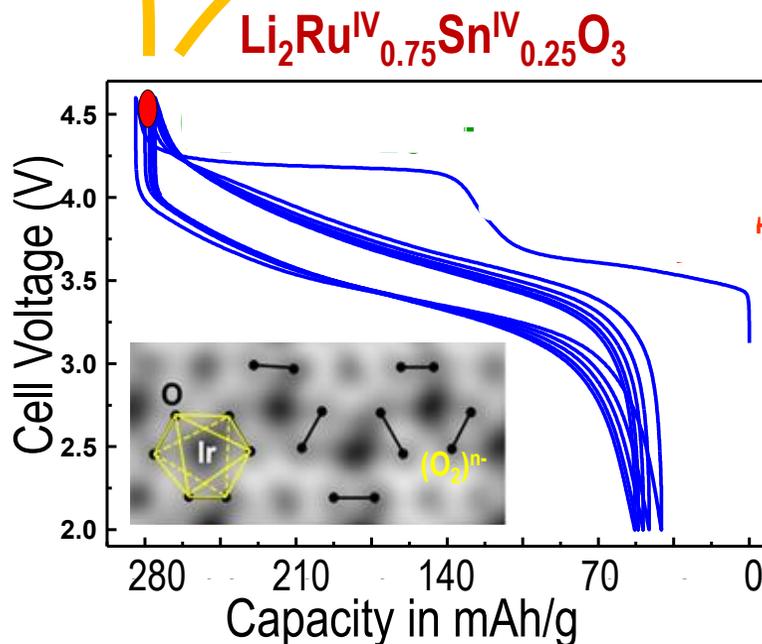
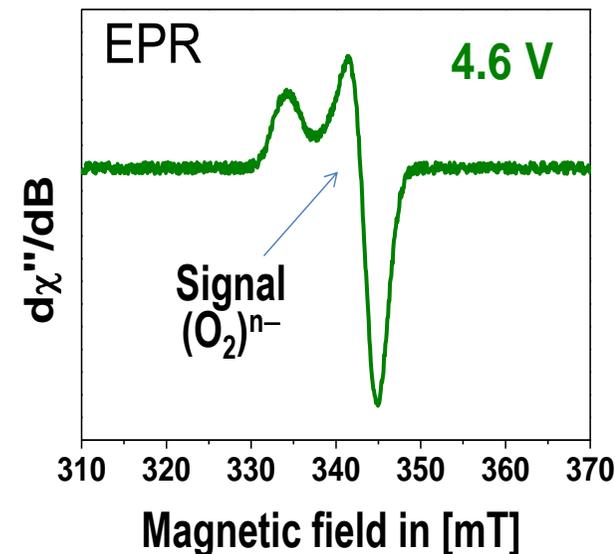


Origin of the extra capacity in Li-rich materials

XPS measurements



EPR measurements

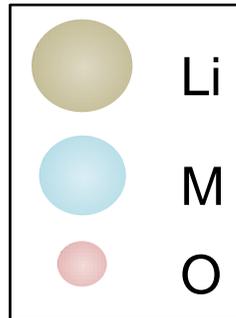
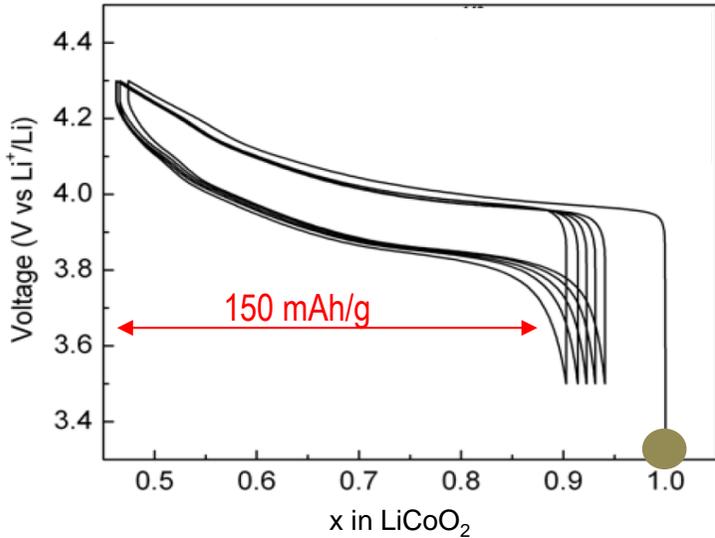
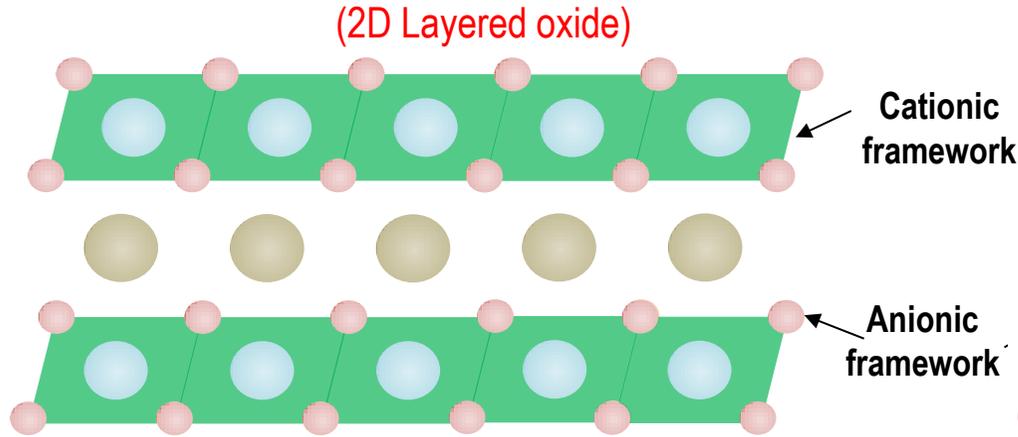


Direct evidence for an anionic redox process ($2O^{2-} \rightarrow (O_2)^{n-}$) in Li-rich lamellar compounds



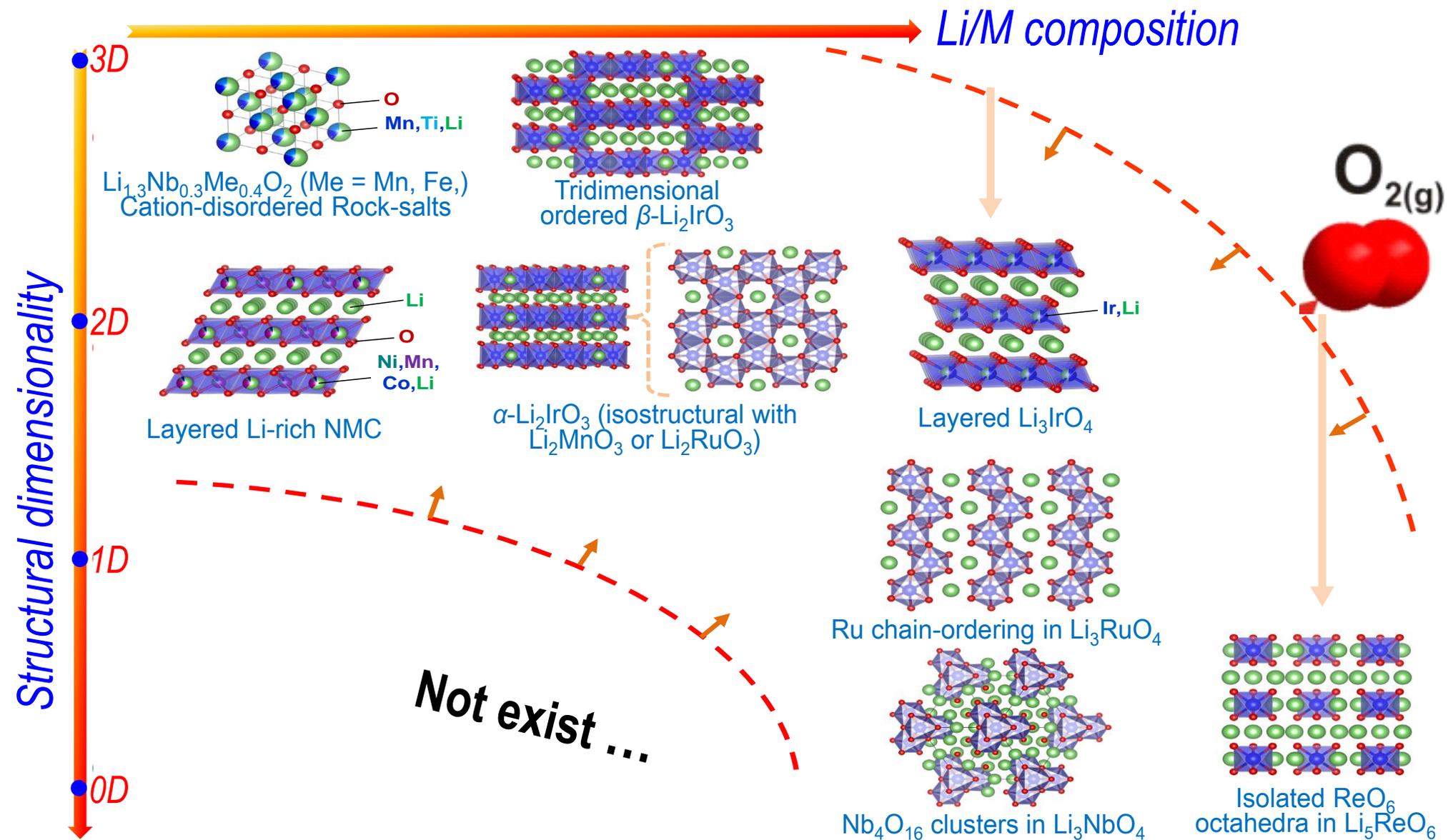
The anionic redox activity: a transformational change

LIB has relied on cationic redox reactions





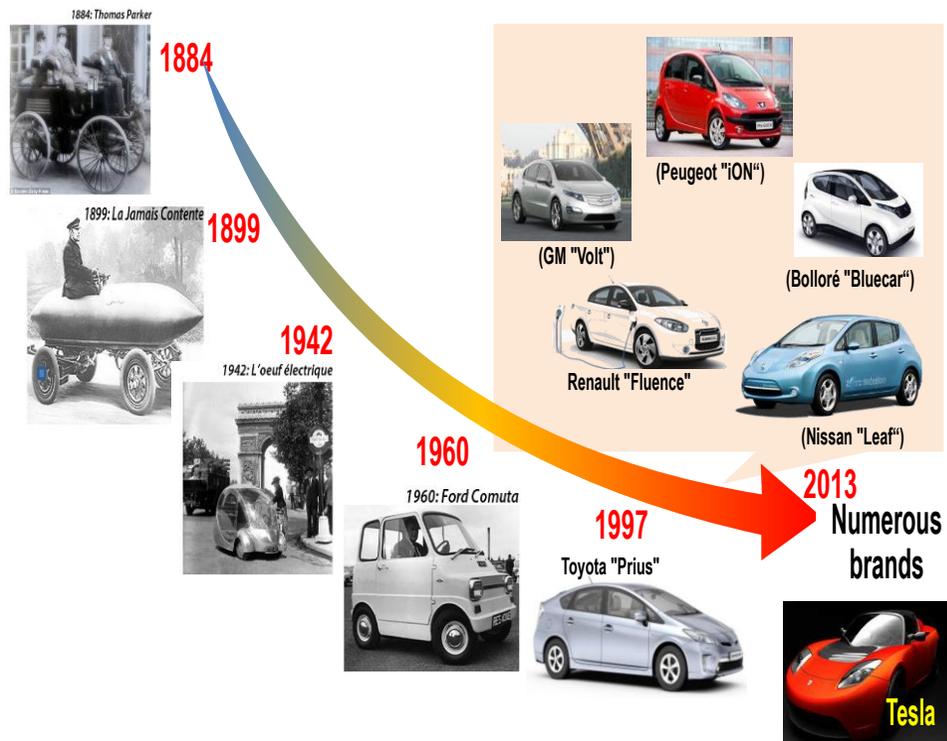
A new playground for designing high capacity electrodes





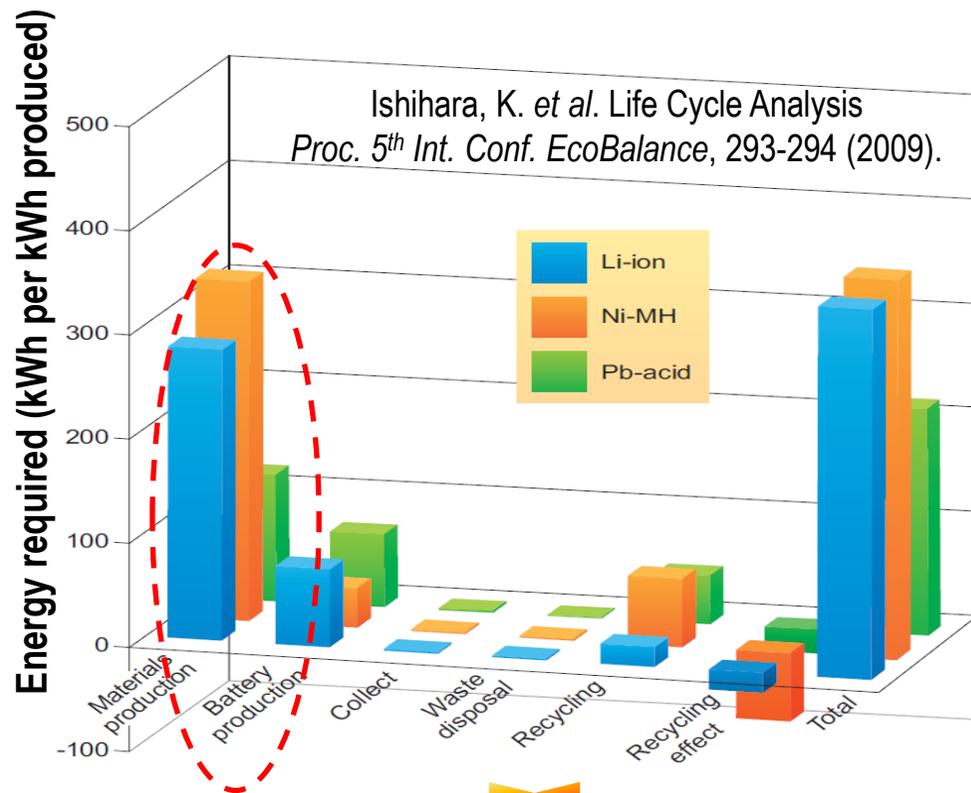
Layered LiMO₂ oxides have enabled today's boom in EV's

Today's EV boom



Is it the perfect solution with respect to sustainability ?

Life cycle analysis



➤ Assembly of a battery of 1kWh

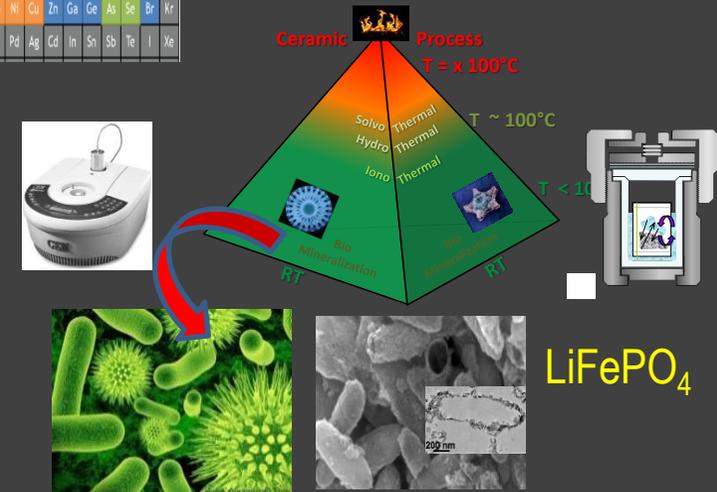
- ✓ Energy needed \approx 327 kWh
- ✓ CO₂ rejected \approx 110 kg



How to make more sustainable Li-ion batteries ? A few trends

- ✓ Design electrode materials based on abundant chemical elements
- ✓ Develop energy-saving synthesis routes
- ✓ Use of renewable organic electrodes
- ✓ Explore chemistries beyond Li

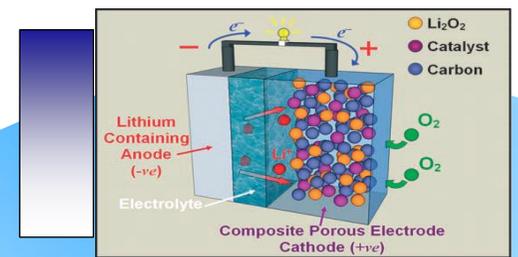
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
H																	He
Li	Be											B	C	N	O	F	Ne
Na	Mg											Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe



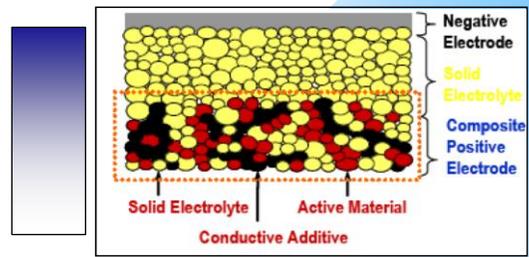


Sustainability thrust: diversification of the battery systems for enhanced performances

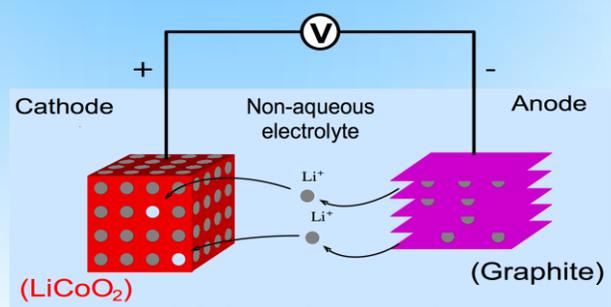
Time to market



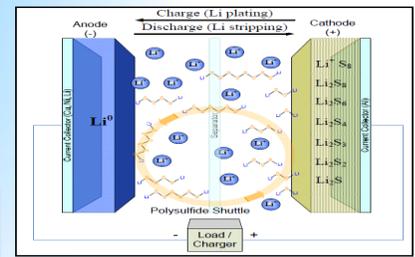
Li/air batteries



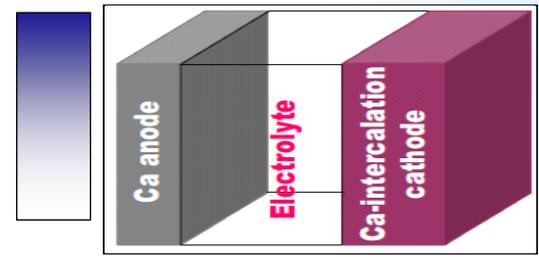
All solid-state (Li/Na) batteries



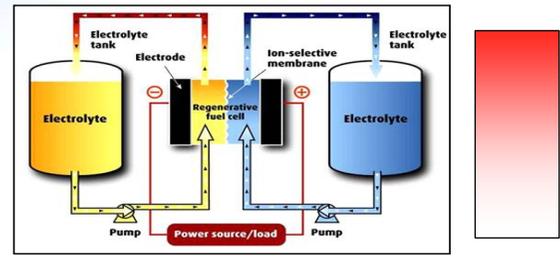
Li-ion batteries



Li/S batteries



Multivalent cation (Mg^{2+} , Ca^{2+}) batteries



Redox flow batteries

None have reached the maturation stage ...

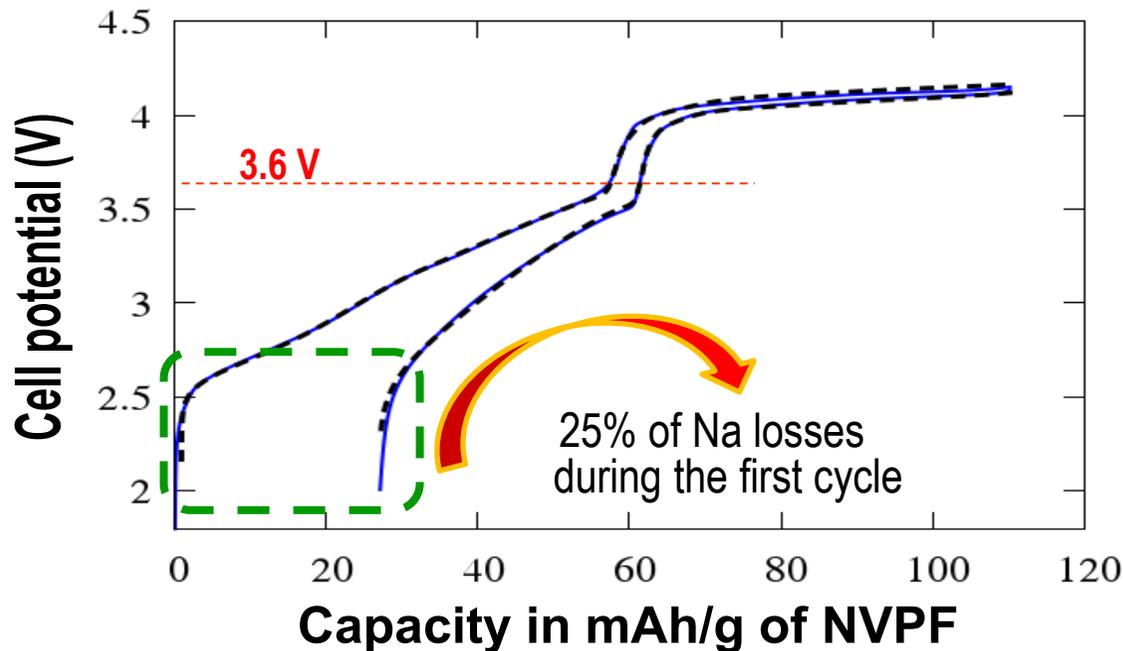
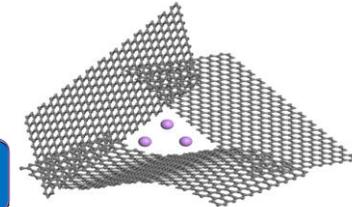
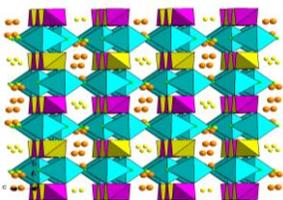


The Na-ion battery: an alternative to Li-ion for sustainability

3
0.98
180.5
0.534
Li
 $1s^2 2s^1$
Lithium

Name	Earth abundance
Hydrogène	0.88 %
Lithium	0.006 %
Béryllium	0.00053 %
Sodium	2.6 %

11
0.9
98
0.97
Na
 $2s^2 2p^6 3s^1$
Sodium



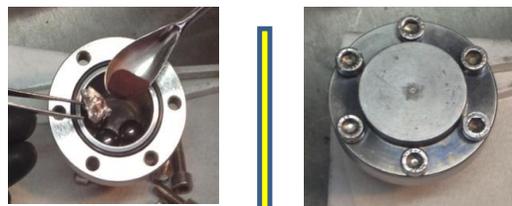
Need of an extra Na source ..

Oversodiated materials
 $(\text{Na}_{3+x}\text{V}_2(\text{PO}_4)_2\text{F}_3)$
???



How to prepare a Na-rich phase via a sustainable approach?

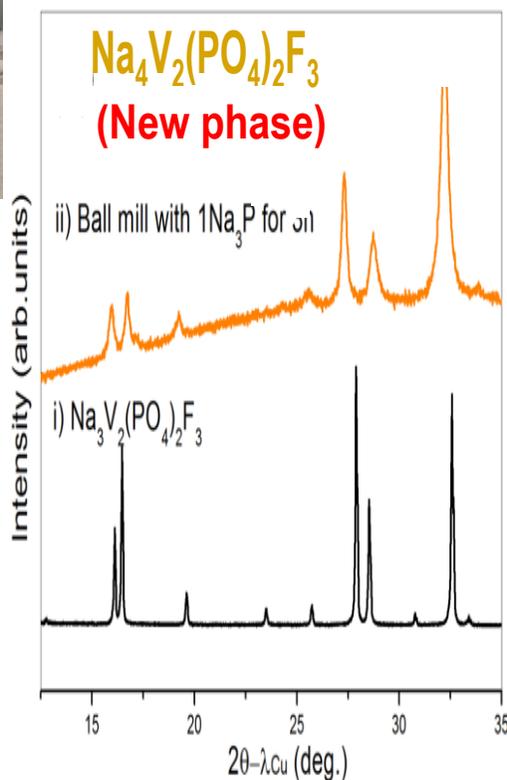
Synthesis via ball milling



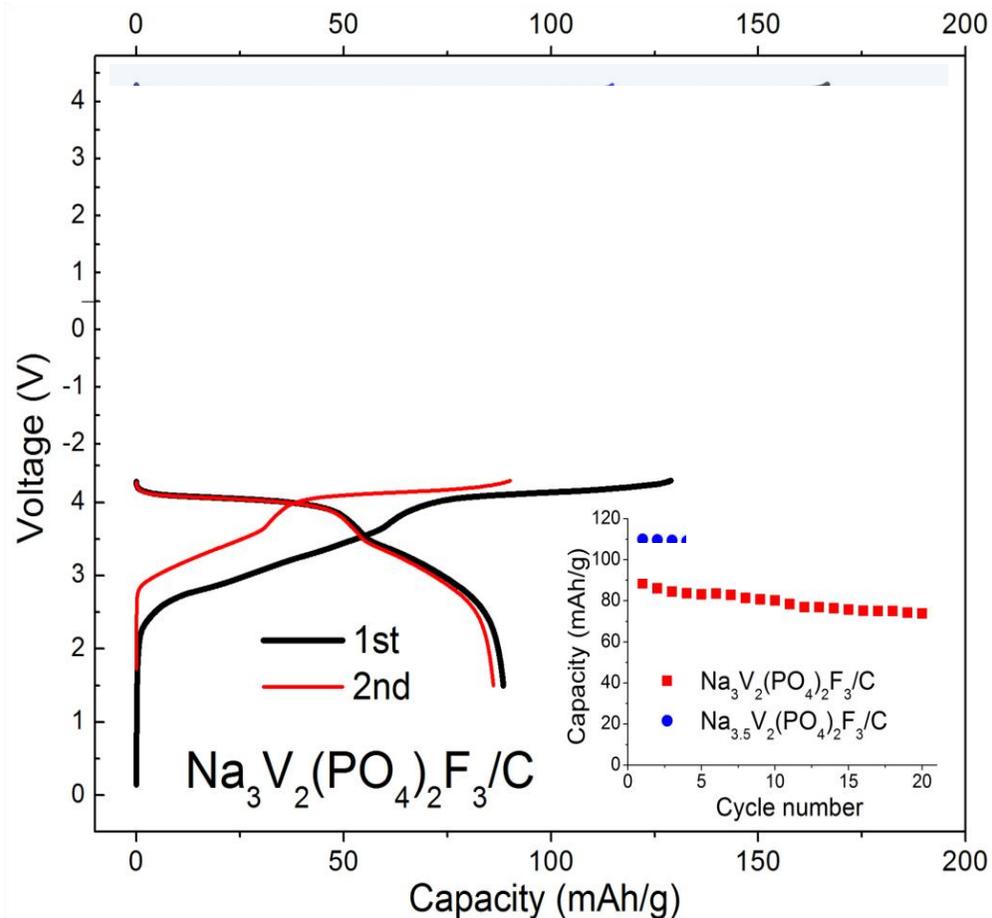
Milling time
= 2 hours



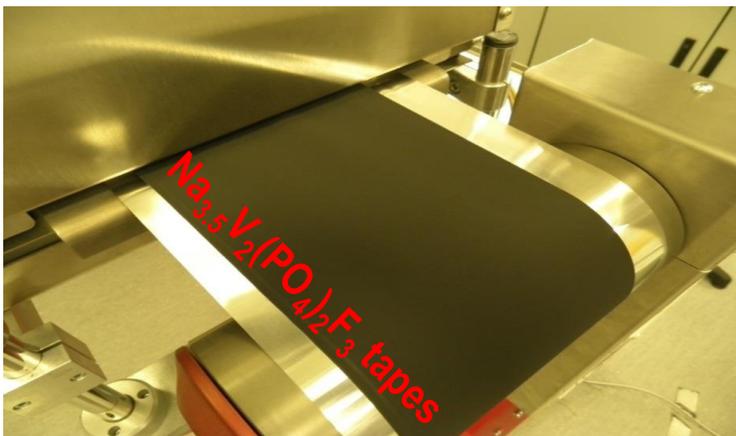
" $\text{Na}_{3+x}\text{V}_2(\text{PO}_4)_2\text{F}_3$
composites
($3 < x < 4$)



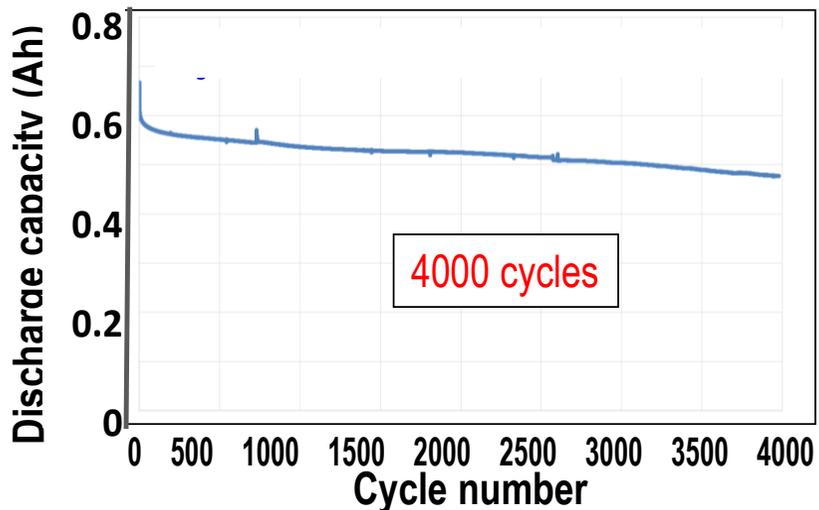
Electrochemical performance



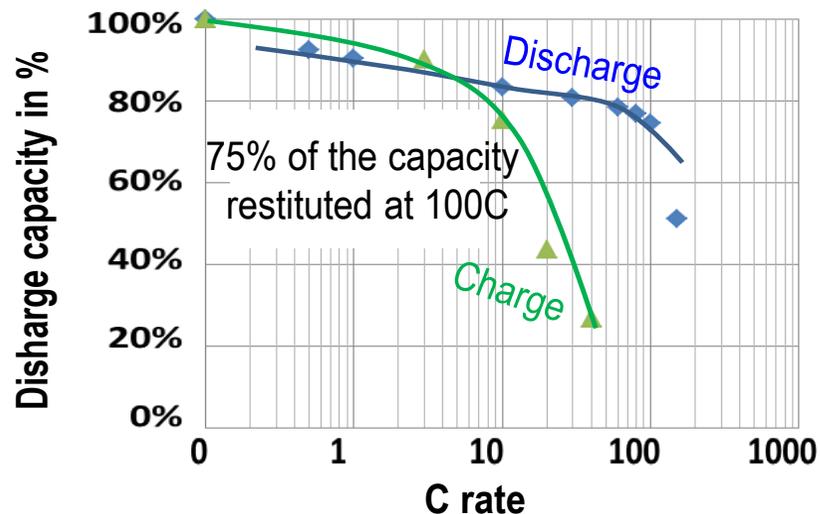
Assembly of 18650 Na-ion cells for benchmarking against Li-ion



Cycle life of C/NVPF Na-ion cells at 1 C



Power rate of C/NVPF Na-ion cells

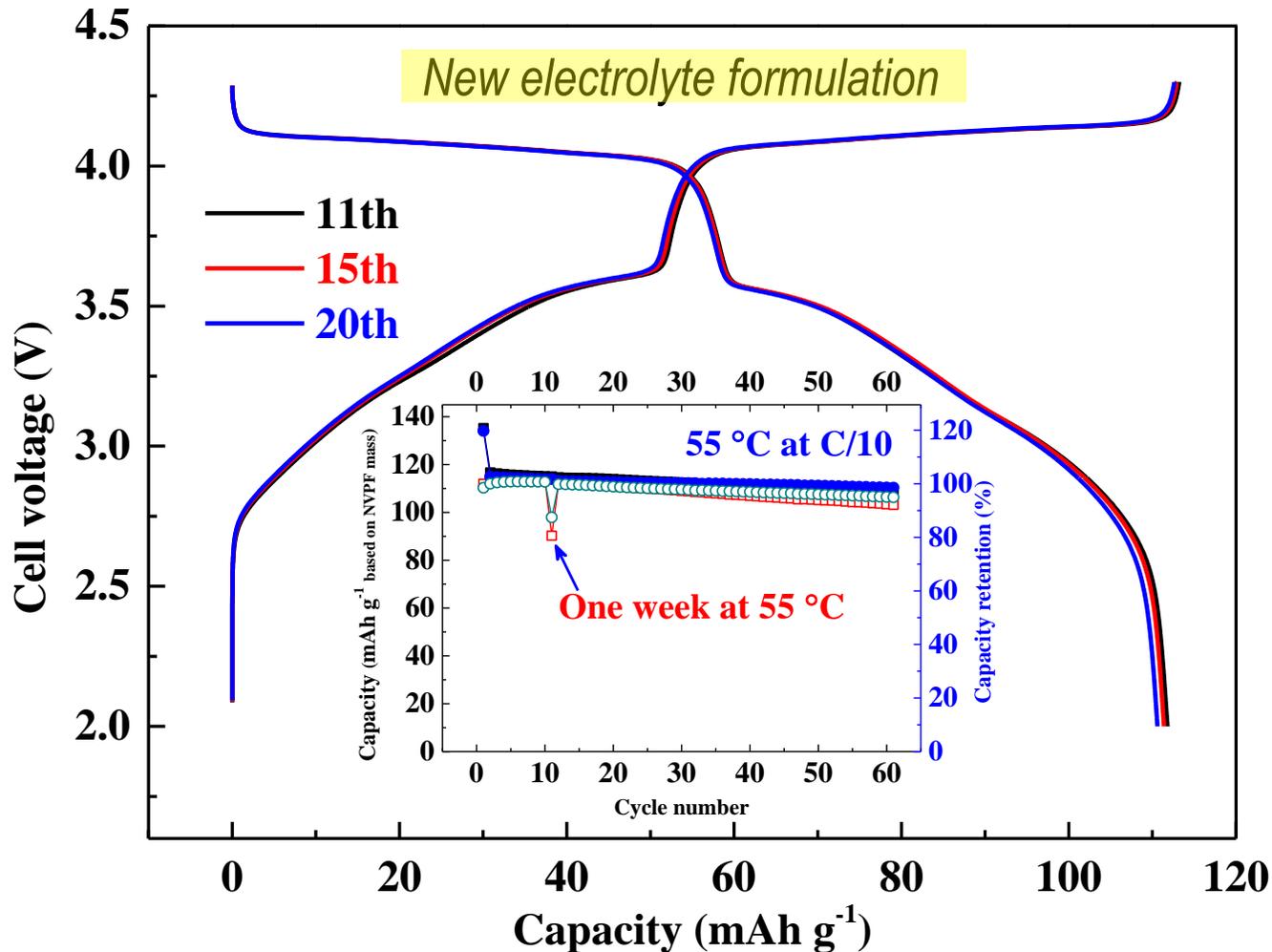


Poor 55°C cycling and self-discharge performances



The $\text{Na}_3\text{V}_2(\text{PO}_4)_2\text{F}_3/\text{C}$ technology: 55°C cycling & self discharge

□ Cycle life & One week self-discharge of NVPF/C at 55°C



TIAMAT



- 105Wh/kg
- ~ 3000 cycles
- 80 % at 10C
- 55°C storage



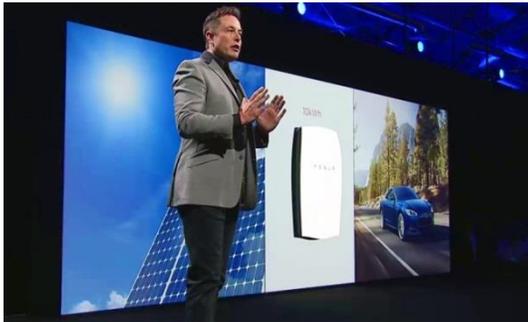
What could be tomorrow's battery research

*Basic problems emerging from
concrete technological challenges*



Looking ahead: New research challenges

Business players change today's traditions



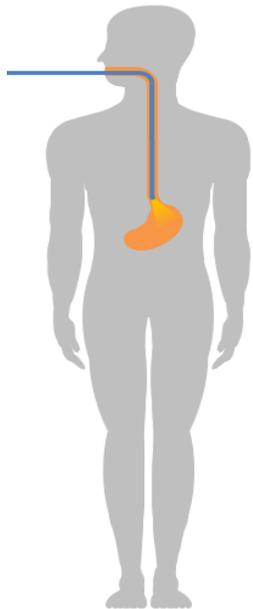
► E. Musk: (100 \$ kWh in 2025)

$$\text{Cost per kWh of stored energy} = \sum \left[\text{Material performance} + \text{Volume production} + \text{Battery second life} \right]$$

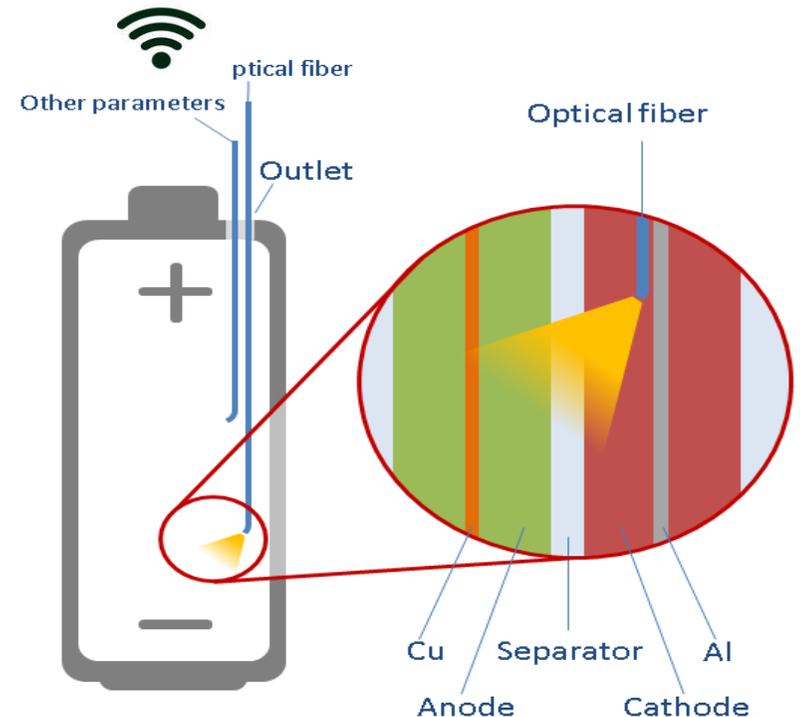
Chemistry/ Material performance Volume production Battery second life

Better traceability

Establish the state of health record of the battery just like for humans



Efforts towards instrumental miniaturization for real time monitoring of the batteries in the field





Looking ahead: New research challenges

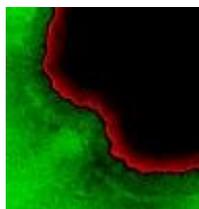
Establish the state of health record of the battery just like for humans



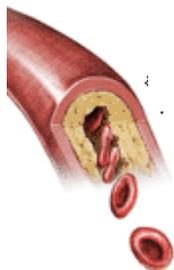
Develop self-healing processes



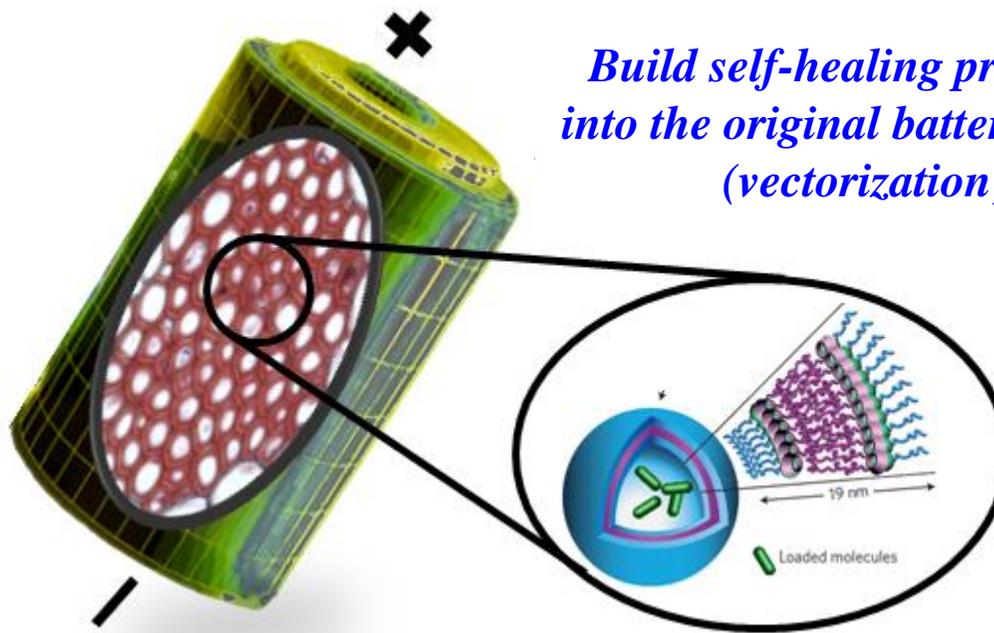
Electrode recovered by an SEI



(Prevents the crossing of Li^+)



Clogged arteria by cholesterol
(Prevents blood circulation)



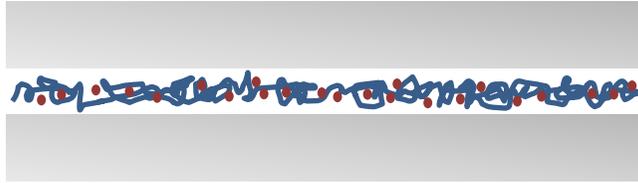
Build self-healing processes into the original battery design (vectorization)

Interdisciplinary research which is still in its embryonic state

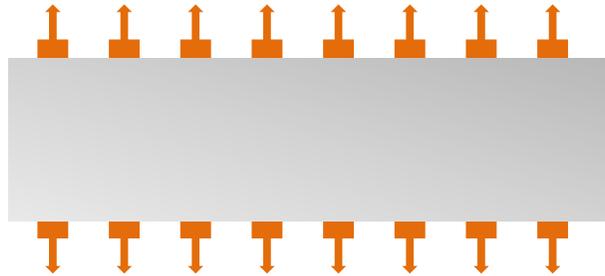


How to tackle this issue ? Innovative chemistry on the battery separator

Multilayers



Surface-
functionalization

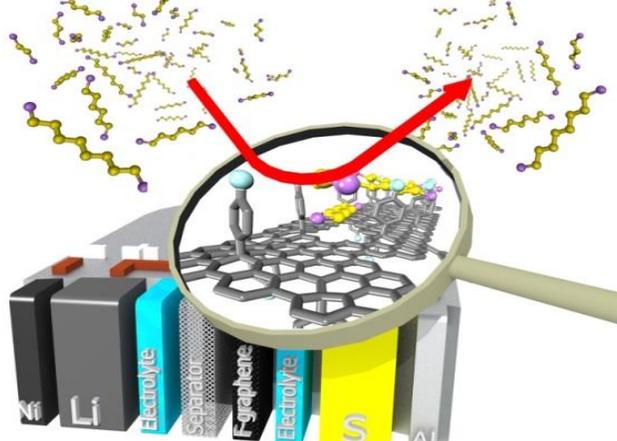


External or internal
stimulation



Functionalization with
 SO_3^- Nafion groups

Fluorinated graphene
oxide

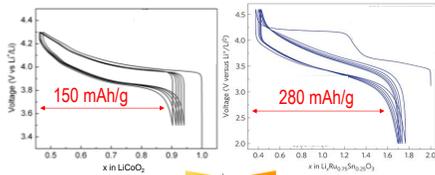


Functionalization to trap species released from side reactions

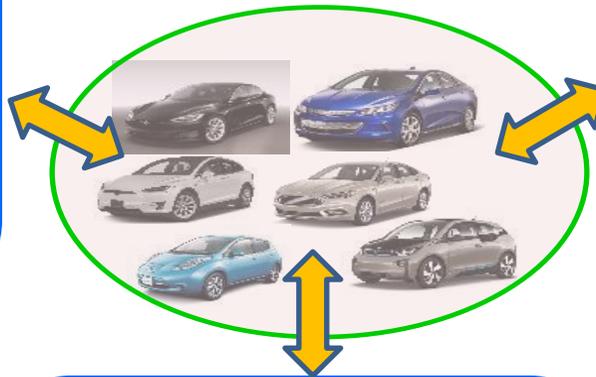


Electric mobility: Conclusions

Anionic redox as a new paradigm to design electrodes with double capacities



Li-ion batteries with ~ 20% energy density improvements



Development of a $C/Na_3V_2(PO_4)_2F_3$ Na-ion technology via materials and electrolytes innovations



Na-ion batteries with attractive power rate performances

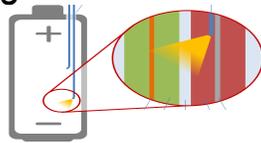
Foreseen EV future

Hardware



Software + services

Future research challenges dealing with SOH batteries



Development of sensing and self-healing processes via innovative chemistry

Foreseen battery future

Materials



Sensing + monitoring)

Acknowledgments



Thank you for your attention

