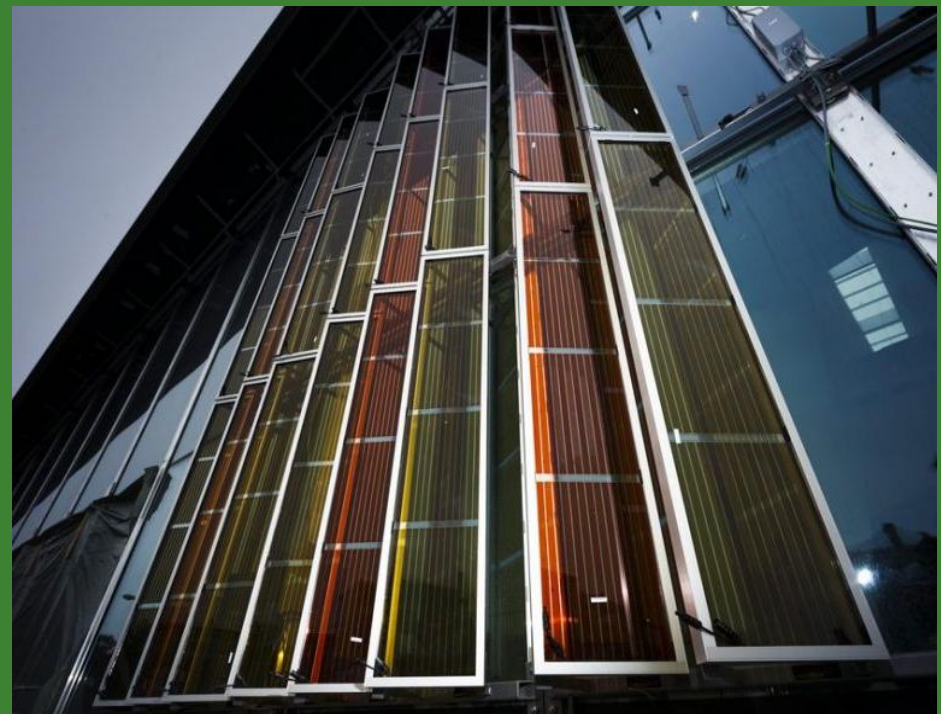
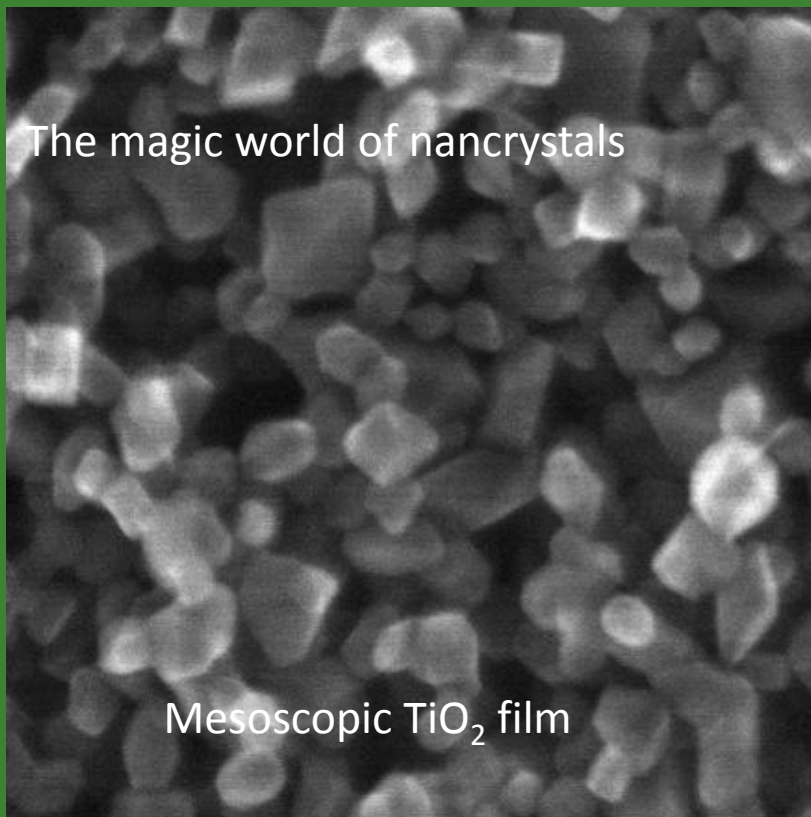


Energy beyond oil:

Mesoscopic photosystems for the generation of fuels from sunlight

Fuel Choices Conference – Panel Discussion Tel Aviv Israel December 4, 2014
The long term future, where is the technology taking us ?

michael.graetzel @epfl.ch



Swiss Tech Convention Center in Lausanne with a glass façade made of dye sensitized solar cells



Solar Energy

Photovoltaics (PV)



Electrical Energy

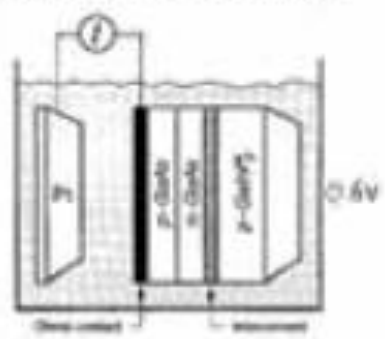
Photoelectrochemical (PEC)

Electrolysis

H_2

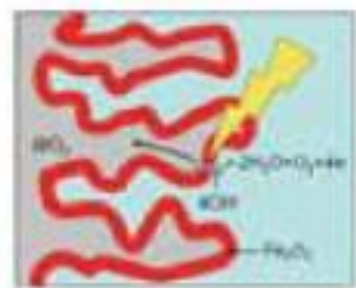
Hydrogen

Monolithic PV+Electrolysis



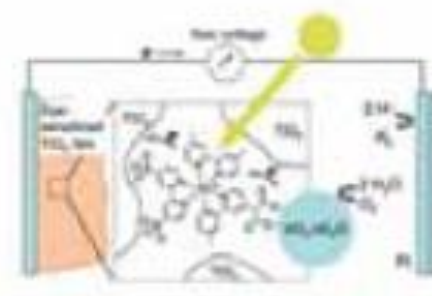
10-15%

Metal Oxide Photoelectrodes

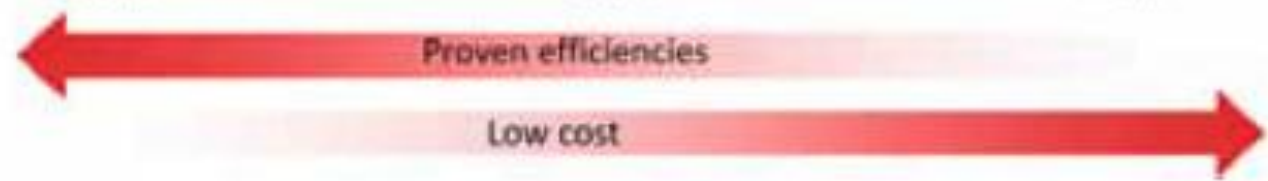


3-5%

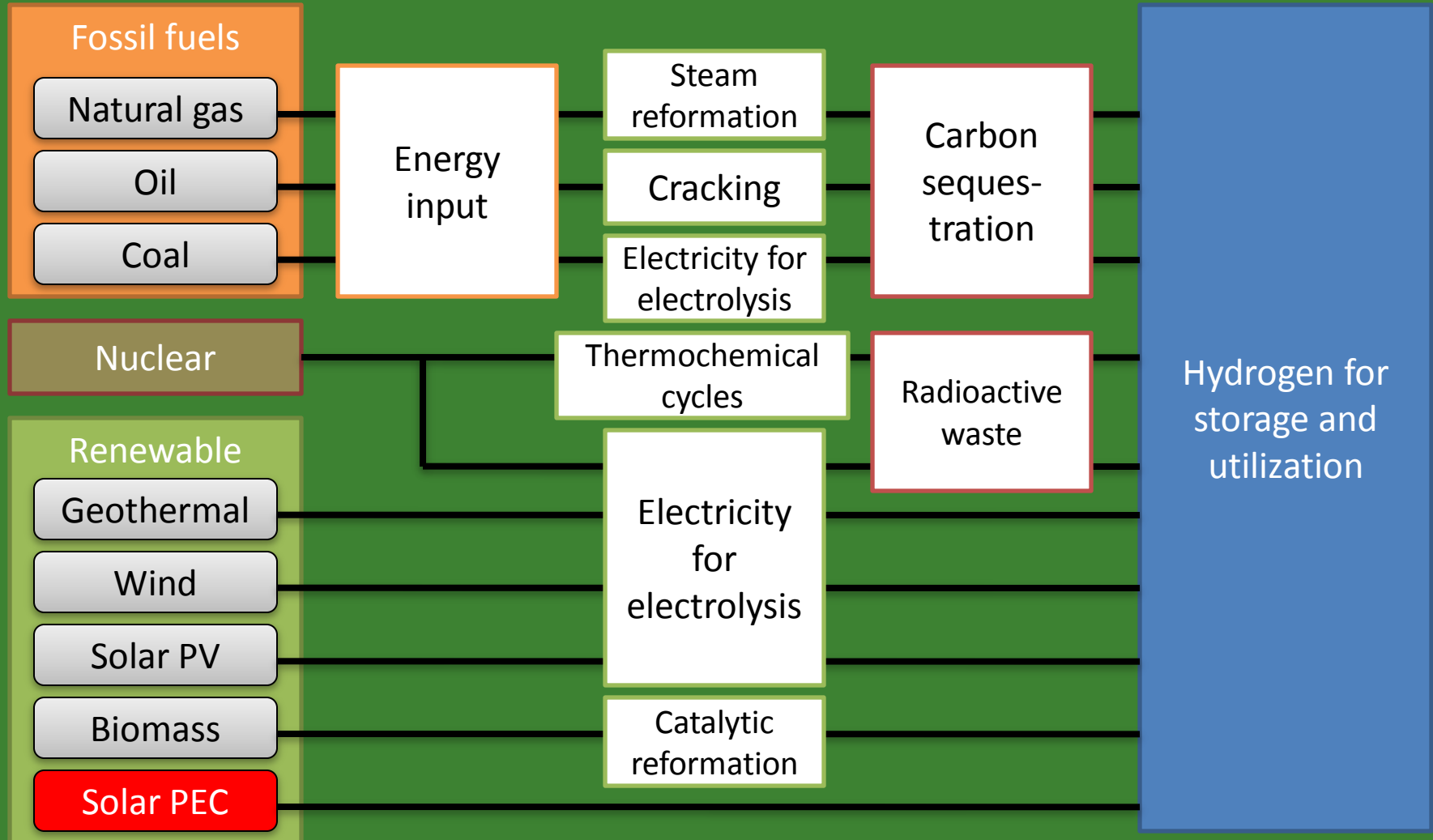
Molecular/Hybrid Systems



< 1%



PEC (photo-electrochemistry) offers a direct path from solar energy to transportation fuels



Source: Toyota Motor Corporation

Our research has introduced mesoscopic structures for energy conversion. These systems are now applied in electrochemical devices for solar generation of chemical fuels, dye sensitized - and perovskite solar cells as well as lithium ions batteries.

ARTICLES

PUBLISHED ONLINE: 7 JULY 2013 | DOI: 10.1038/NMAT3684

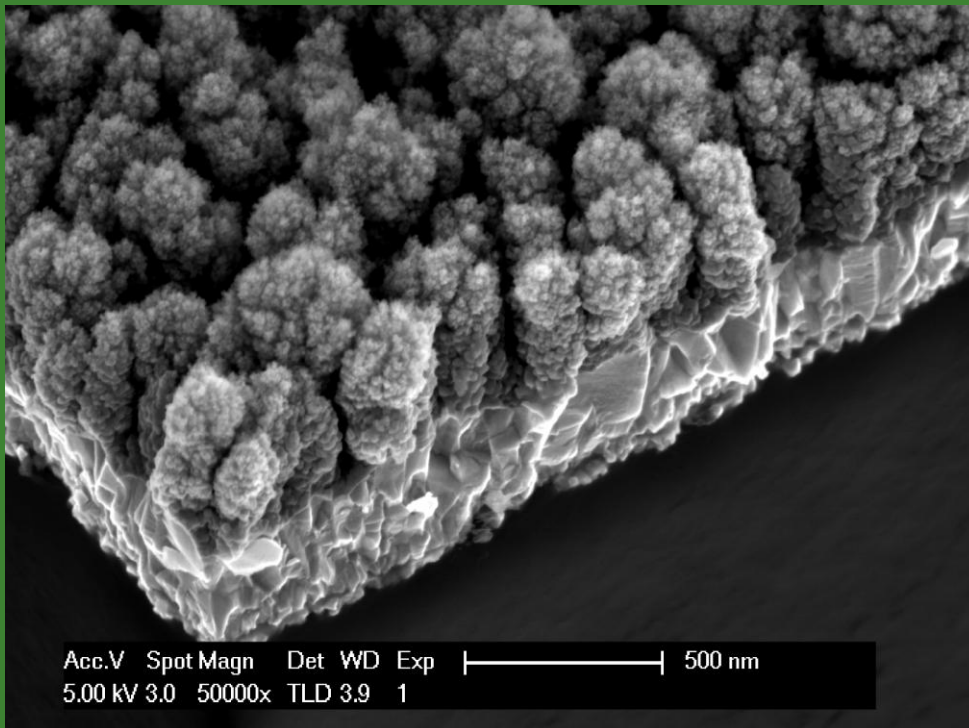
nature
materials

Identifying champion nanostructures for solar water-splitting

Scott C. Warren^{1*}, Kislou Voitchovsky², Hen Dotan³, Celine M. Leroy¹, Maurin Cornuz¹, Francesco Stellacci², Cécile Hébert⁴, Avner Rothschild³ and Michael Grätzel¹

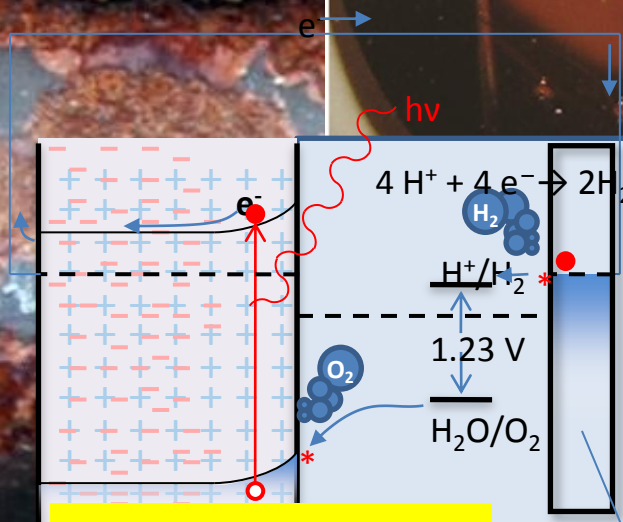
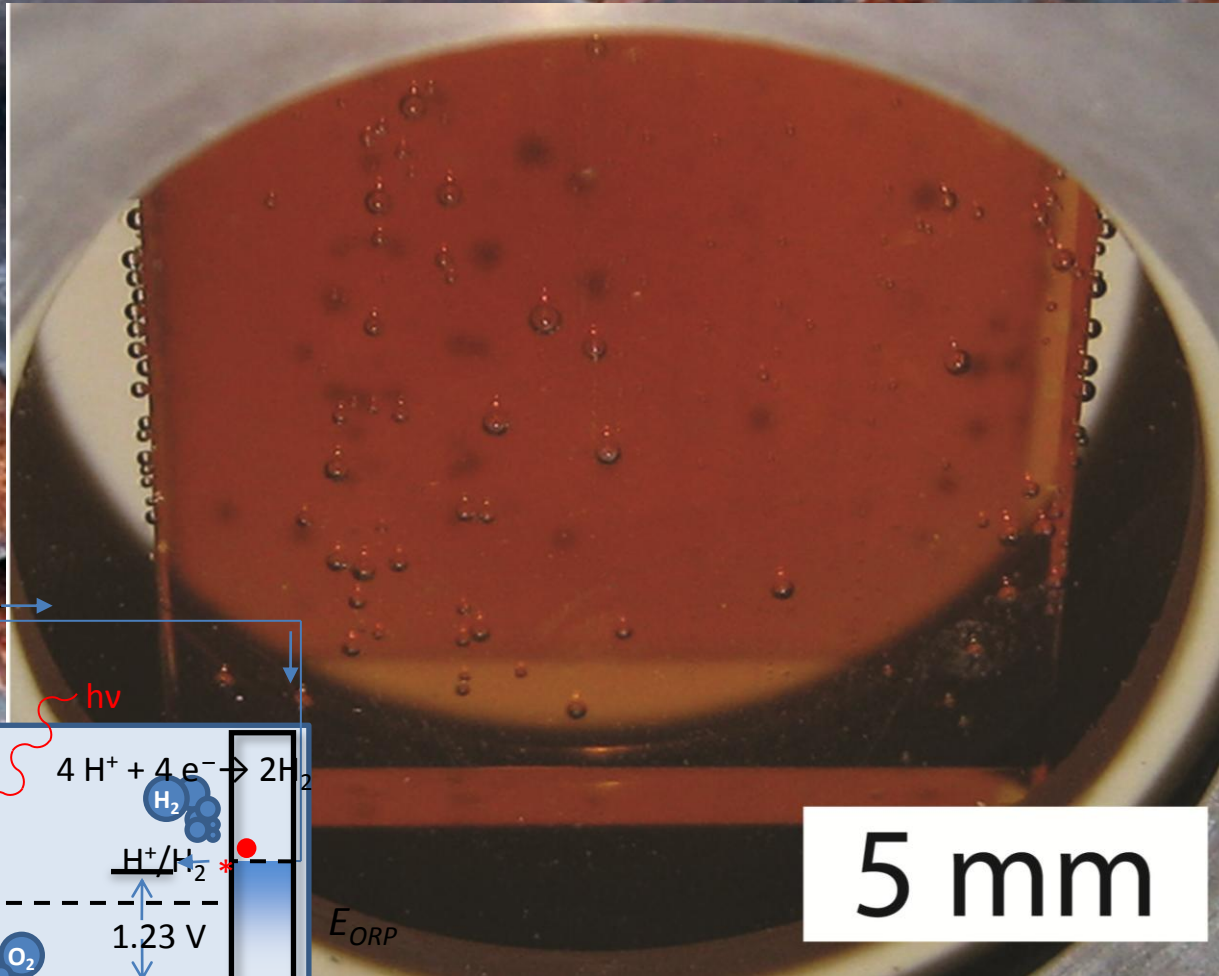
$$\eta_{\text{coll}} = (1 + (d/L_D)^2)^{-1}$$

$$\eta_{\text{coll}} = 1/(1 + \tau_{\text{trans}}/\tau_{\text{rec}})$$



The key advantage of mesoscopic over planar semiconductor architectures is that they can **achieve near quantitative collection of the photo-generated charge carriers** even for materials where the charge-carrier diffusion length is much shorter than the light absorption length. In photo-electrochemical cells **the electrons and holes are used to generate hydrogen and oxygen from water,**

The power of the mesoscopic architecture; visible light induced oxygen evolution on Fe₂O₃ (iron oxide ,rust)



5 mm

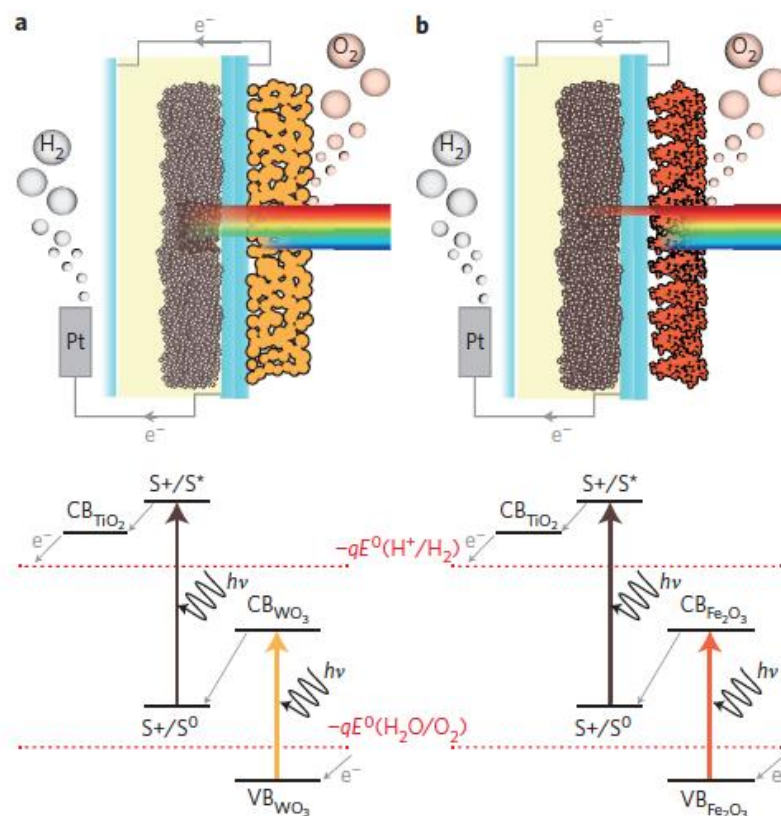
Rust is inactive !

n-type Semiconductor / Aqueous electrolyte / Metal cathode
 Net Reaction: $2\text{H}_2\text{O} + h\nu \rightarrow 2\text{H}_2 + \text{O}_2$

Highly efficient water splitting by a dual-absorber tandem cell

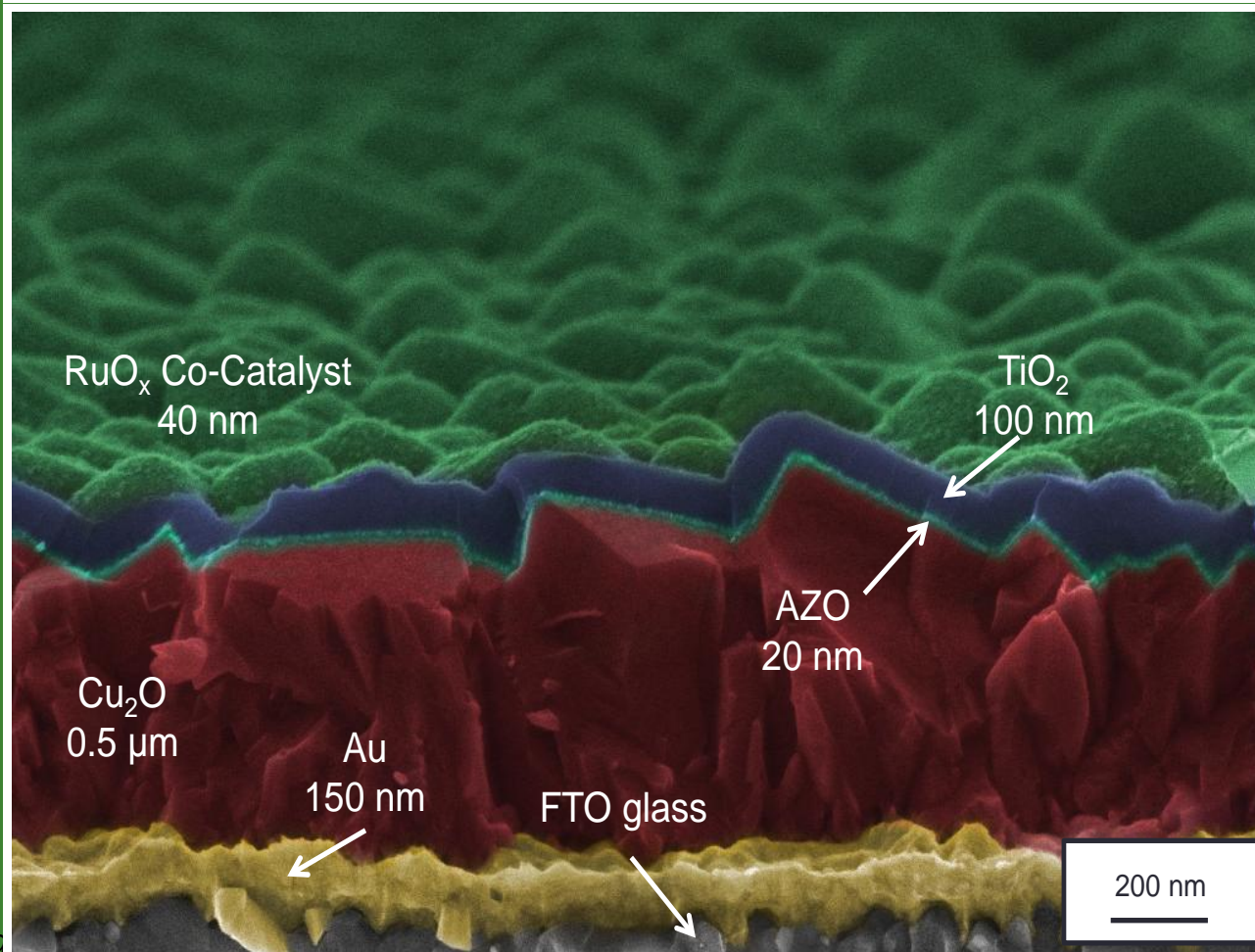
Jeremie Brillet¹, Jun-Ho Yum¹, Maurin Cornuz¹, Takashi Hisatomi¹, Renata Solarska², Jan Augustynski², Michael Graetzel¹ and Kevin Sivula^{1*}

Photoelectrochemical water-splitting devices, which use solar energy to convert water into hydrogen and oxygen, have been investigated for decades. Multijunction designs are most efficient, as they can absorb enough solar energy and provide sufficient free energy for water cleavage. However, a balance exists between device complexity, cost and efficiency. Water splitters fabricated using triple-junction amorphous silicon^{1,2} or III-V³ semiconductors have demonstrated reasonable efficiencies, but at high cost and high device complexity. Simpler approaches using oxide-based semiconductors in a dual-absorber tandem approach^{4,5} have reported solar-to-hydrogen (STH) conversion efficiencies only up to 0.3% (ref. 4). Here, we present a device based on an oxide photoanode and a dye-sensitized solar cell, which performs unassisted water splitting with an efficiency of up to 3.1% STH. The design relies on carefully selected redox mediators for the dye-sensitized solar cell^{6,7} and surface passivation techniques⁸ and catalysts⁹ for the oxide-based photoanodes.



Highly active oxide photocathode for photoelectrochemical water reduction

Adriana Paracchino¹, Vincent Laporte², Kevin Sivula¹, Michael Grätzel¹ and Elijah Thimsen¹*†



Build-in Cu₂O/ZnO (AZO) p/n junction and TiO₂ overlayer protect the Cu₂O film Against photo-corrosion

The Holy Grail of photocatalysis: solar splitting of water into hydrogen and oxygen on Cu_2O films protected by TiO_2



REPORT

Science **345**, 1593 (2014)

Science

AAAS

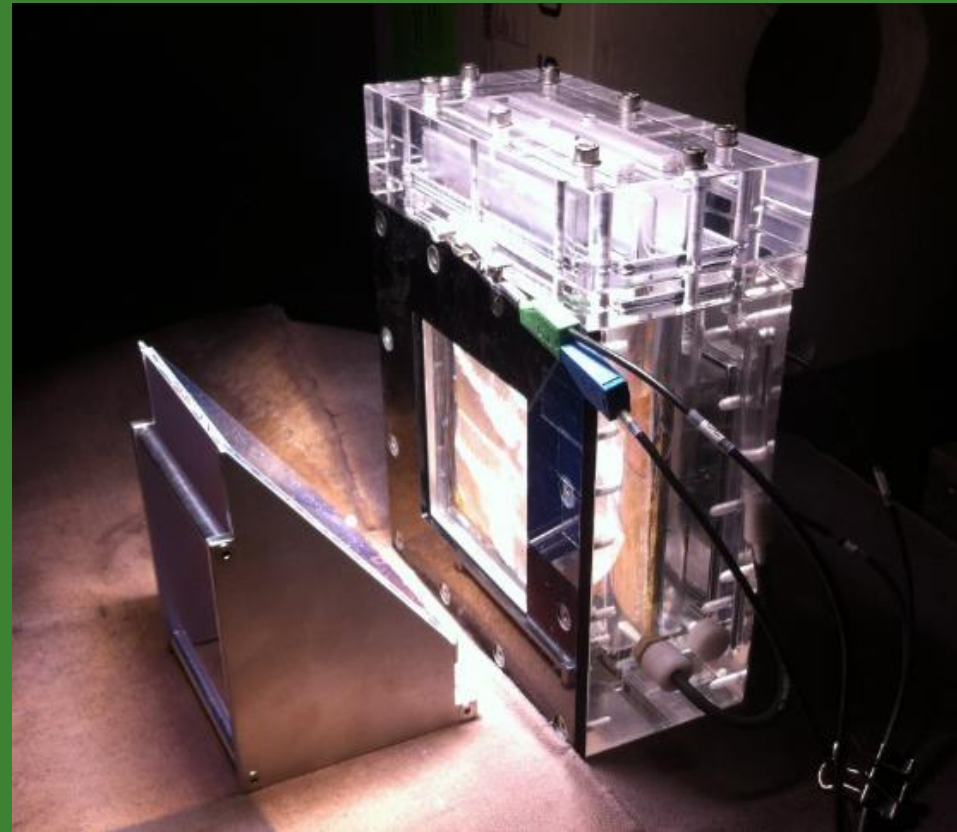
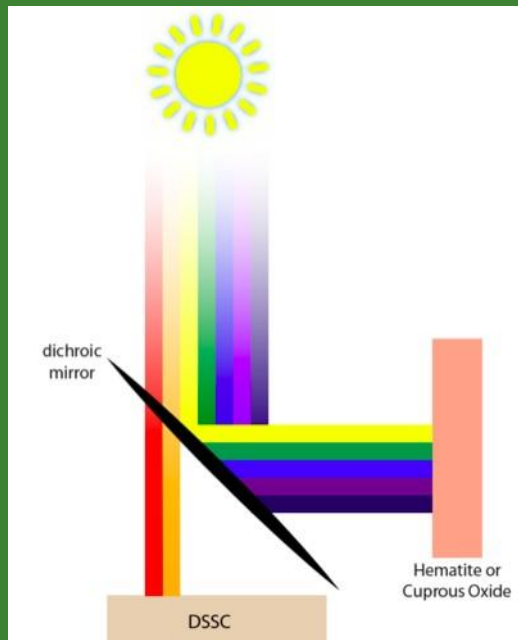
WATER SPLITTING

Water photolysis at 12.3% efficiency via perovskite photovoltaics and Earth-abundant catalysts

Jingshan Luo,^{1,2} Jeong-Hyeok Im,^{1,3} Matthew T. Mayer,¹ Marcel Schreier,¹
Mohammad Khaja Nazeeruddin,¹ Nam-Gyu Park,³ S. David Tilley,¹
Hong Jin Fan,² Michael Grätzel^{1*}

Present status of research and development

- Large electrode tandem (100 cm²)
- 10% unassisted water splitting in AM1.5 sunlight.



Future practical implementation: PECDEMO -EU project sunlight to hydrogen

Fuel Cells and Hydrogen Joint Undertaking (FCH JU)

- Stand-alone hybrid PEC-PV devices will be constructed and tested for performance and stability,
- fabrication of large scale demonstration modules that will be tested in the field under real-world conditions.
- Price target for hydrogen: 5€/kg

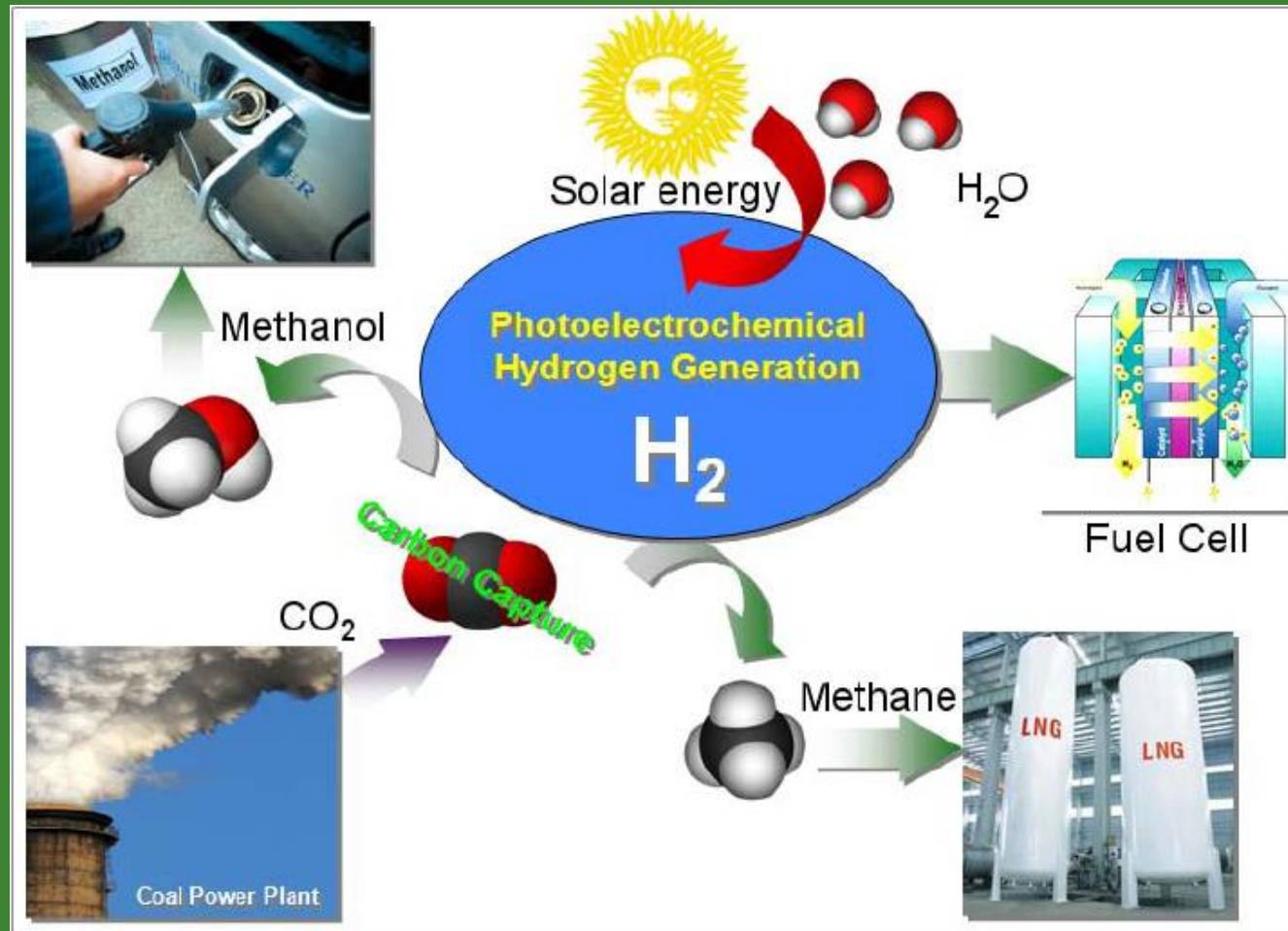


Prof. Roel van de Krol

Participant no.	Participant organisation name	Country
1 (Coordinator)	Helmholtz-Zentrum Berlin für Materialien und Energie GmbH Prof. Roel van de Krol	Germany
2	Ecole Polytechnique Fédérale de Lausanne Prof. Michael Grätzel	Switzerland
3	Technion – Israel Institute of Technology Prof. Avner Rothschild	Israel
4	Deutsches Zentrum für Luft - und Raumfahrt Dr. Christian Jung Dr. Martin Roeb	Germany
5	Universidade do Porto Prof. Adélio Mendes	Portugal
6	Evonik Industries AG Dr. Matthias Blug	Germany
7	Solaronix SA Dr. Toby Meyer	Switzerland

H₂ is a key future energy vector and fuel

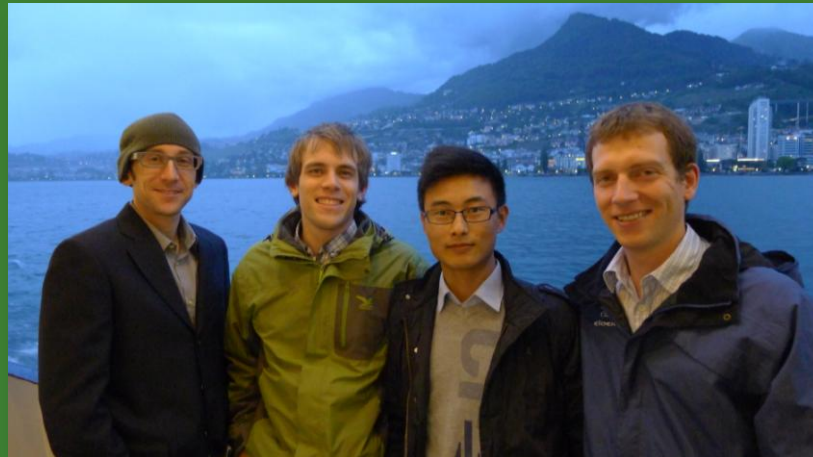
- Provides environmental, economic and national security
- Can be easily converted to methane or methanol
- Price target for renewable hydrogen 5 €/kg
- By 2050, H₂ is expected to comprise 50% of transportation fuels.





LPI-SolarFuels group members

- Ludmilla Steier Fe_2O_3 photo-anodes for solar water splitting
- David Tilley Group leader, now Professor at the University of Zurich
- Marcel Schreier CO_2 reduction on photocathodes, gas chromatography
- Jingshan Luo copper oxide-and CIGS based photocathodes, tandems, earth-abundant electrocatalysts
- Matt Mayer present solar fuels group leader, photocathodes, metal oxides, mesoscopic photoelectrodes
- Min-Kyu Son Cu_2O tandems, scale-up of water splitting systems



Cooperations with Israeli groups

Academic

- Professor Efrat Lifshitz Technion Haifa
- Professor Avner Rothschild Technion Haifa
- Dr. Hen Dotan Technion Haifa
- Professor Doron Aurbach Bar-Ilan University
- Professor Ariel Zaban Bar-Ilan University
- Professor Lioz Etgar Hebrew University
- Professor Gary Hodes Weizman Institute
- Professor David Cahen Weizman Institute

Industrial

3GSolar Jerusalem

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Swiss Federal Office of Energy SFOE



We thank our corporate sponsors

